

Aquatic Plant Extracts

Crops

Identification of Petitioned Substance

Chemical Names:

Not applicable

11 Acadian™ Organic Powder 0.5-0.3-14 (OMRI,
12 2005)**Other Names:**

Kelp extracts

CAS Numbers:84775-78-0 (Extracts of *Ascophyllum nodosum*)**Trade Names:**Alg-A-Mic
Stress-X Powder**Other Codes:**

None

Characterization of Petitioned Substance

Composition of the Substance:

Aquatic plant extracts are composed of chemicals naturally found in aquatic plants, as well as breakdown products that are formed during the manufacturing process. Common sea plants identified in the literature as sources of aquatic plant extracts include knotted wrack (*Ascophyllum nodosum*) and Sea Bamboo (*Ecklonia maxima*) (NCR, 2004). Aquatic plant extracts used as soil amendments are most commonly derived from kelp, specifically from *Ascophyllum* spp. and other seaweed harvested from the North Atlantic (Baldwin, 2001).

Specific chemical ingredients are not listed in product information available for most commercial aquatic plant extract products. However, aquatic plants contain proteins, lipids, sugars, amino acids, and nutrients, vitamins, plant hormones, and other biochemicals. A laboratory analysis of liquid aquatic plant extracts, provided by Laboratoires GOEMAR (undated) as an attachment to the Original TAP Database Form (1995), listed the following substances as components of aquatic plant extracts:

- **Amino Acids:** glutamic acid, alanine, aspartic acid, methionine, glycine, ethanolamine, lysine, cystine, threonine, serine, praline, phenylalanine, valine, leucine, isoleucine, argenine, tyrosine,
- **Vitamins:** carotene, fucoxanthin, thiamine (B1), riboflavin (B2), pantothenic acid, niacin, B12, C, D3, E, K
- **Phytohormones:** glycosyl cytokinins, auxins and gibberellins
- **Other components:** polysaccharides (i.e., complex sugars), betaines¹

Kelp contains a wide range of naturally occurring plant nutrients and trace minerals essential to plant growth, health, and productivity. Aquatic plant extracts also contain carbohydrates (e.g., alginic acid and mannitol) that bind micronutrients and help make them more available to crop plants. These components of aquatic plant extracts also contribute to building soil structure (Acadian Seaplants Limited, Undated).

Cytokinins, a class of plant hormones present in aquatic plant extracts, have been reported to have beneficial effects on crops, including increase in number or size of fruits or seed heads, synchronization of flowering within a field, and delayed decay of mature plants (Baldwin, 2001; Baker, 1996). When used as a plant dip, cytokinins

¹ Many plants use betaines to regulate and maintain osmotic pressure in their cells. This allows the plant to achieve tolerance to water shortages or heavy salt stresses (Nishimura et al., 2001).

49 may also reduce transplant stress (Hall, 1997). Although other plant hormones (e.g., auxins, gibberellins) are
50 present in aquatic plant extracts (Allen et al., 2001; Baker, 1996), they are not likely to be present in concentrations
51 sufficient to have noticeable effect on plant growth (Baldwin, 2001). For this reason, some marketed products
52 may be enhanced with synthetic hormones to ensure performance of the product (Baldwin, 2001).

53

54 **Properties of the Substance:**

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56 Aquatic plant extracts are described as a brownish-black powder or brown liquid with a mild marine (or
57 fish) odor (Original TAP Database Form, 1995; Necessary Organics, 1991). The material is soluble in water
58 and ethanol, is stable, and is not subject to hazardous polymerization (Original TAP Database Form, 1995;
59 Necessary Organics, 1991). It is non-flammable (Necessary Organics, 1991).

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61 **Specific Uses of the Substance:**

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63 According to the Original TAP Database Form (1995), aquatic plant extracts are used as foliar fertilizers (on
64 all crops) or as soil conditioners. Aquatic plant extracts also are used in combinations as a foliar/soil feed
65 or transplant solution and seed treatment. The material is then absorbed into the plant and acts as a
66 growth promoter. Aquatic plant extracts can be used as a source of micronutrients and growth promoters;
67 as a rooting solution for transplants and cuttings; for cold hardiness in tomato plant, citrus fruits, and
68 cabbage; and to reduce pest damage (nematodes in tomato and okra; mites in strawberries, peaches and
69 apples) (Original TAP Database Form, 1995).

70

71 Aquatic plant extracts are sometimes applied as a foliar spray by farmers seeking a natural source of
72 micronutrients. For the most part, none of the micronutrient levels in kelp extracts are high enough to
73 correct a deficiency, but they are used as a "tonic" providing a broad array of micronutrients and other
74 trace elements in organic farming (Hall and Sullivan, 2001). Some aquatic plant extract products are
75 supplemented with synthetic nutrients (Hall and Sullivan, 2001).

76

77 **Approved Legal Uses of the Substance:**

78

79 National Organics Program (NOP)

80

81 Aquatic plant extracts (other than hydrolyzed) are currently on the National List for use as a plant or soil
82 amendment. Annotations to the National List limit the allowable extraction process to those using
83 potassium hydroxide or sodium hydroxide solvents, and the solvent amount used is limited to that amount
84 necessary for extraction. Extraction processes are described further in Evaluation Question #1.

85

86 EPA

87

88 Extracts of *Ascophyllum nodosum* (CAS number 84775-78-0) are on EPA's list 4B of inert ingredients in
89 pesticides.

90

91 EPA exempted cytokinin in aqueous extract of seaweed meal (as opposed to seaweed extract) from the
92 requirement for a tolerance in all food commodities when used as a plant growth regulator on plants,
93 seeds, or cuttings and on all food commodities after harvest (EPA, 1998).

94

95 **Action of the Substance:**

96

97 As previously mentioned, aquatic plant extracts are used as foliar fertilizers or as soil conditioners. An
98 understanding of the mechanisms of action that produce the beneficial effects of aquatic plant extracts is
99 incomplete (Henry, 2005). Research has demonstrated that many crops do not respond to seaweed extract
100 when soil fertility and production conditions are favorable (Kuisma, 1989; McGeary and Birkenhead, 1984).
101 Other studies have shown that crops are more likely to respond to kelp products under stressful growing
102 conditions (Verkleij, 1992; Nelson and Van Staden, 1984; Abetz, 1980). However, a study in 1995 found no

103 beneficial effects to tomatoes of kelp-derived foliar sprays when the plants were inadvertently subjected to
104 stress early in growth (Tourte, 1997).

105
106 Foliar fertilization is subject to variables associated with (1) uncertainty and variability of nutrient content
107 of material sprayed and (2) leaf uptake processes (Tourte, 1997). More specifically, the product efficacy
108 may be influenced by the particular species used to make the extract, the growth stage of the plants, the
109 time of harvest and post-harvest handling of the plants used in the product, the processing method, and
110 the product's shipment and storage conditions (Henry 2005). Effectiveness of kelp extracts probably varies
111 with crop type, quality of the foliar spray itself, soil nutrient availability, timing of spray applications, and
112 soil moisture and weather conditions. For some crop and/or conditions, foliar treatment with kelp
113 products may have no effect (Tourte, 1997).

114
115 Aquatic plant extracts used as a soil amendment improve soil fertility by making nutrients and hormones
116 more directly available to plants. Aquatic plant extracts contain carbohydrates such as alginic acid and
117 mannitol, which enable plants to better absorb nutrients from the soil. The carbohydrates are broken down
118 resulting in the stimulation of beneficial soil bacteria that fix nitrogen making it available to plant roots.
119 Furthermore, these activities decrease the need for chemical fertilizers (Smittle, 1991).

120
121 Additionally, hormones (e.g., auxins, cytokins, and gibberellins) contained in plant extracts are absorbed
122 into the plant and act as growth promoters. When used as a plant growth regulator, seaweed extract
123 would be expected to alter cell division, root and shoot elongation, initiation of flowering, and other
124 metabolic functions (in contrast to fertilizers, which simply supply minerals needed for nutrition) (Allen et
125 al., 2001).

126

127 **Status**

128

129 **International**

130 **Canada** - Canadian General Standards Board (CGSB) -
131 http://www.pwgsc.gc.ca/cgsb/032_310/32.310epat.pdf

132

133 As of June 2004 draft (CGSB, 2004), aquatic plant extracts were accepted for use in Canada if they were
134 natural (nonsynthetic). The only synthetic solvents allowed for extraction are potassium hydroxide or
135 sodium hydroxide, which are not to exceed the amount necessary for extraction. If the product contains
136 other synthetic preservatives (e.g., formaldehyde) or is fortified with otherwise prohibited plant nutrients,
137 then the aquatic plant product is prohibited in Canada.

138

139 **CODEX Alimentarius Commission** - <ftp://ftp.fao.org/docrep/fao/005/Y2772e/Y2772e.pdf>

140

141 Based on Table 1 of the CODEX, substances for use in soil fertilizing and conditioning - seaweeds and
142 seaweed products - need recognition by the certification body or authority.

143

144 **European Economic Community (EEC) Council Regulation 2092/91 -**

145 http://europa.eu.int/eur-lex/en/consleg/pdf/1991/en_1991R2092_do_001.pdf

146

147 Aquatic plant extracts are allowed in the EU, following Annex IIB - Seaweed and seaweed products
148 (Organic Trade Association, 2002).

149

150 Seaweeds and seaweed products are allowed for use as fertilizers and soil conditions, as far as directly
151 obtained by:

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- 153 • Physical processes including dehydration, freezing and grinding;
- 154 • Extraction with water or aqueous acid and/or alkaline solution; and
- 155 • Fermentation;

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157 Need recognition by the inspection body or inspection authority.

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Japan Agricultural Standard for Organic Production –
<http://www.ams.usda.gov/nop/NOP/TradeIssues/JAS.html>

No information pertaining to aquatic plant extracts was located at this site.

California Certified Organic Growers International (CCOF) –
<http://www.ccof.org/pdf/GlobalMarketAccessDraftForReview.pdf>

No information pertaining to aquatic plant extracts was located at this site.

Washington State Department of Agriculture: European Organic Verification Program (EOVP) –
<http://agr.wa.gov/FoodAnimal/Organic/default.htm>

No information pertaining to aquatic plant extracts was located at this site.

Evaluation Questions for Substances to be used in Organic Crop or Livestock Production

**Evaluation Question #1: Is the petitioned substance formulated or manufactured by a chemical process?
(From 7 U.S.C. § 6502 (21))**

Alkali Extraction of Aquatic Plant Extracts:

According to the Original TAP Database Form (1995), seaweed extract is produced from fresh, live plants which are processed into a soluble powder or liquid, and may be stabilized with synthetic acids and fortified with other ingredients.

Aquatic plant extract manufacturers utilize an alkali extraction process to “digest” the plants and derive both micronutrients and naturally occurring plant hormones from plant sources. This process also transforms the plants into a soluble, easily transported form. The majority of manufacturers use potassium hydroxide as the primary reagent in the alkali extraction process. Other alkali reagents used by some manufactures include sodium hydroxide, calcium hydroxide, and sodium carbonate (Henry, 2005).

Alkali extraction may be conducted either at ambient or elevated temperatures and pressures. Some manufacturers claim that there is a reduced yield of desired plant growth hormones when the products are extracted at higher temperatures. However, no studies have been found comparing the effectiveness of extracts produced at different concentrations of alkali, or at different temperatures (Henry, 2005).

Following alkali digestion, most manufacturers reduce the pH of the product with an acid, most commonly phosphoric acid (Henry, 2005). Phosphoric acid is a synthetic material (Original TAP Review, 1999). Only the amount of acid needed to achieve a pH of 3.5 is recommended (Conrad, 2002). It is not clear if neutralization is a necessary step. One form of the most widely-used extract product (Maxicrop) is not neutralized, and a 25 percent solution has a pH of 9-10. No studies have been found comparing the effectiveness of high-pH versus neutralized alkali extracts (Henry, 2005).

Nonsynthetic Production of Aquatic Plant Extracts

According to the Original TAP Database Form (1995), aquatic plant extracts can also be derived naturally. For example, kelp can be dehydrated after harvest by sun drying, and then ground into a meal product. This kelp meal can be sprinkled directly on the soil, or diluted with water and either sprayed on plant foliage as a foliar spray or poured directly into the ground as a soil drench (3R Lighting, 2005).

Nonsynthetic products also may be produced using mechanical disruption, or freezing, pulverization, and clarification of the thawed slurry (Henry, 2005). The relative efficacy of alkali-extracted versus non-alkali-extracted product has not been consistently demonstrated, perhaps partly as a result of a lack of

212 understanding of the mechanism by which aquatic plant extracts exert any purported beneficial effect
213 (Henry, 2005).

214
215 **Evaluation Question #2: Is the petitioned substance formulated or manufactured by a process that**
216 **chemically changes the substance extracted from naturally occurring plant, animal, or mineral sources?**
217 **(From 7 U.S.C. § 6502 (21).)**

218
219 Yes, some chemical reactions do occur during the processes by which aquatic plant extracts are obtained
220 from plant sources, such as kelp and seaweed. However, because aquatic plant extracts are complex
221 chemical mixtures, it may not be possible to characterize all the reactions and the extent to which there are
222 chemical changes beyond the simple effects of shifting pH.

223
224 Alkali “digestion” of aquatic plants involves the breakdown of galactans in seaweed cell walls (Sideman,
225 2004). Galactans are tough, fibrous polymeric saccharides (primarily repeating units of galactose). With
226 alkali extraction, this breakdown is accomplished through partial hydrolysis of the galactans, catalyzed by
227 bases (such as sodium hydroxide, potassium hydroxide, potassium carbonate) used in sufficient quantity
228 to achieve a pH of 9 - 10 (Sideman, 2004). Without alkali extraction this breakdown can be accomplished
229 through non-chemical means such as freeze and thaw or grinding.

230
231 **Evaluation Question #3: Is the petitioned substance created by naturally occurring biological**
232 **processes? (From 7 U.S.C. § 6502 (21).)**

233
234 Aquatic plant extracts contain substances that naturally occur in aquatic plants or that are produced when
235 aquatic plants are broken down. As described in Evaluation Question #1, aquatic plant extracts can be
236 produced using nonsynthetic methods. However, aquatic plant extracts produced by alkali digestion, as
237 currently permitted by the National List, are not derived with a naturally occurring biological process.

238
239 **Evaluation Question #4: Is there environmental contamination during the petitioned substance’s**
240 **manufacture, use, misuse, or disposal? (From 7 U.S.C. § 6518 (m) (3).)**

241
242 According to Tooby (2003) plant extracts appear to be biodegradable and are likely to have a low impact on
243 crops or stored agricultural products. Additionally, there is no information available from EPA to suggest
244 that environmental contamination results from the manufacture, use, misuse, or disposal of aquatic plant
245 extracts. Environmental contamination could result from improper disposal of acids or bases used in alkali
246 extraction.

247
248 **Evaluation Question #5: Is the petitioned substance harmful to the environment? (From 7 U.S.C. § 6517**
249 **(c) (1) (A) (i) and 7 U.S.C. § 6517 (c) (2) (A) (i).)**

250
251 No information was uncovered to suggest that aquatic plant extracts could cause toxicity to plants, soil
252 organisms, or higher animals (Original TAP Database Form, 1995).

253
254 The potential for over-harvesting of kelp/seaweed fields for production of aquatic plant extracts may be an
255 environmental concern (Chalker-Scott, date unknown; Original TAP Database Form, 1995). However, no
256 evidence has been found indicating that this potential impact is actually occurring.

257
258 Additionally, nutrients (i.e., nitrogen, phosphorous, and potassium) found in runoff from excessively or
259 improperly applied fertilizers can cause excess algae growth in surface water (i.e., eutrophication). Excess
260 algae can, in turn, use up oxygen in the water, potentially harming fish and other aquatic animals. No
261 information has been found indicating that aquatic plant extracts are responsible for excessive nutrient
262 runoff.

263

264 **Evaluation Question #6: Is there potential for the petitioned substance to cause detrimental chemical**
265 **interaction with other substances used in organic crop or livestock production? (From 7 U.S.C. § 6518**
266 **(m) (1).)**

267
268 Based on the intended use of the substance, no information was uncovered to suggest that aquatic plant
269 extracts could cause detrimental chemical interaction with other substances used in organic crop
270 production.

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272 **Evaluation Question #7: Are there adverse biological or chemical interactions in the**
273 **agro-ecosystem by using the petitioned substance? (From 7 U.S.C. § 6518 (m) (5).)**

274
275 Some aquatic plant extracts produced by sodium hydroxide extraction may contain excess sodium. If this
276 product is applied at high concentrations, plants may suffer (e.g., foliar damage, decreased plant growth)
277 due to their sensitivity to salt (Original TAP Database Form, 1995).

278
279 Aquatic plant extract products that are not neutralized may have a high pH, which has a potential to
280 adversely affect plants. However, no evidence of aquatic plant extract products adversely affecting crops
281 has been identified. No studies have been found comparing the effectiveness of high-pH versus
282 neutralized alkali extracts (Henry, 2005).

283
284 **Evaluation Question #8: Are there detrimental physiological effects on soil organisms, crops, or**
285 **livestock by using the petitioned substance? (From 7 U.S.C. § 6518 (m) (5).)**

286
287 As previously mentioned, aquatic plant extracts can be used as a source of micronutrients and growth
288 promoters and potash²; as a rooting solution for transplants and cuttings; for cold hardiness in tomato
289 plant, citrus fruits, and cabbage; and to reduce pest damage (including nematodes in tomato and okra;
290 mites in strawberries, peaches and apples) (Original TAP Database Form, 1995).

291
292 Aquatic plant extracts are unlikely to create unacceptable changes in soil temperature, water availability,
293 pH levels, nutrient availability, or salt concentration. However, if aquatic plant extracts produced by
294 sodium hydroxide extraction (which may contain excess sodium) are applied at high concentrations, the
295 plants may suffer from foliar damage (burn) due to their sensitivity to salt (Original TAP Database Form,
296 1995). Additionally, as previously stated, aquatic plant extract products that are not neutralized may have
297 a high pH, and this has the potential to affect plants adversely. For example, micronutrients (i.e., copper,
298 iron, manganese, zinc, and chloride) tend to be less available in soils with high pH.

299
300 **Evaluation Question #9: Is there a toxic or other adverse action of the petitioned substance or its**
301 **breakdown products? (From 7 U.S.C. § 6518 (m) (2).)**

302
303 Based on their intended use, aquatic plant extracts and their breakdown products are unlikely to have toxic
304 or other adverse actions. Aquatic plant extracts are composed primarily of natural plant constituents, such
305 as proteins, lipids, glucides, amino acids, and vitamins (Original TAP Database Form, 1995). These
306 components are generally biodegradable and nontoxic, and many can be utilized by organisms.

307
308 **Evaluation Question #10: Is there undesirable persistence or concentration of the petitioned substance**
309 **or its breakdown products in the environment? (From 7 U.S.C. § 6518 (m) (2).)**

310
311 According to Tooby (2003), aquatic plant extracts appear to be biodegradable. Thus, the products and their
312 breakdown products are unlikely to have an undesirable persistence in the environment.

313

² Potash refers to a variety of mined and manufactured salts, all containing the element potassium in water-soluble form (USGS, 2001).

314 **Evaluation Question #11: Is there any harmful effect on human health by using the petitioned**
315 **substance? (From 7 U.S.C. § 6517 (c) (1) (A) (i), 7 U.S.C. § 6517 (c) (2) (A) (i) and 7 U.S.C. § 6518 (m) (4).)**
316

317 Other than the potential for lung irritation due to inhalation of the soluble powder form of aquatic plant
318 extract, no adverse effects on human health from the intended use were identified. According to the
319 Material Safety Data Sheet (MSDS) for Necessary Organics' SeaMix Liquid Fertilizer, presented as an
320 attachment to the Original 1995 TAP Database Form, the liquid form poses "no known health hazards"
321 (Original TAP Database Form, 1995). According to Tooby (2003), plant extracts are believed to be
322 associated with likely low adverse effects on human health.
323

324 However, the potential for sensitivity or intolerance to seaweed and, therefore, aquatic plant extracts, may
325 exist. For example, it may be possible for some people to be sensitive to monosodium glutamate (MSG)³,
326 which is found naturally in protein-rich food products, including seaweed. There are some reports that
327 mild, temporary reactions to MSG may occur in a small portion of the population. Nonetheless, MSG is
328 not considered an allergen by the American College of Allergy, Asthma and Immunology (IFIC, 1997).
329 Additionally, the U.S. Food and Drug Administration has found no evidence to suggest any long-term,
330 serious health consequences from consuming MSG (IFIC, 1997).
331

332 **Evaluation Question #12: Is there a wholly natural product which could be substituted for the**
333 **petitioned substance? (From 7 U.S.C. § 6517 (c) (1) (A) (ii).)**
334

335 Natural products that may substitute for aquatic plant extract in its intended use include nonsynthetic
336 fertilizers and soil amendments, including manure, blood meal, bone meal, compost, feather meal, guano,
337 and other nonsynthetic animal or plant products. Also, aquatic plant extracts produced by nonsynthetic
338 means (as described in Evaluation Question #1) are potential substitutes for synthetically produced
339 products.
340

341 Although manure is a complete fertilizer, it lacks vital nutrients. An NPK of 1-1-1 is typical for manure⁴,
342 though the nutrient content of manure is dependent upon the diet and species of the animal that produced
343 it. Fresh manure contains the highest level of nutrients needed for plant growth.
344

345 **Evaluation Question #13: Are there other already allowed substances that could be substituted for the**
346 **petitioned substance? (From 7 U.S.C. § 6518 (m) (6).)**
347

348 According to the NOP rule (7 CFR 205.601(j)), other allowed plant or soil amendments that could be
349 substitutes for aquatic plant extracts, based on the intended uses, include the following:
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- 351 • **Humic acids** - naturally occurring deposits, water and alkali extracts only;
- 352
- 353 • **Micronutrients** - not to be used as a defoliant, herbicide, or desiccant. Those made from nitrates or
354 chlorides are not allowed. Soil deficiency must be documented by testing; and
355
- 356 • **Liquid fish products** - can be pH adjusted with sulfuric, citric or phosphoric acid. The amount of
357 acid used shall not exceed the minimum needed to lower the pH to 3.5; (NOP, 2005).
358

359 **Evaluation Question #14: Are there alternative practices that would make the use of the petitioned**
360 **substance unnecessary? (From 7 U.S.C. § 6518 (m) (6).)**
361

362 Potential alternative practices include recycling compost or other sources of nutrients. Alternative soil
363 building practices include using cover crops, compost and/or manures, reducing tillage, avoiding traffic
364 on wet soils, and maintaining soil cover with plants and/or mulches (Feenstra et al., 1997).
365

³ MSG is a sodium salt of the amino acid glutamate (IFIC, 1997).

⁴ See: <http://www.ext.vt.edu/departments/envirohort/factsheets2/fertilizer/jan89pr6.html>

366 REFERENCE LIST

- 367
- 368 Abetz, P. 1980. Seaweed extracts: Have they a place in Australian agriculture or horticulture? *The Journal of*
369 *the Australian Institute of Agricultural Science* 46:23-29 (as cited in Tourte, 1997).
- 370
- 371 Acadian Seaplants Limited. Undated. ASL™ kelp meal for soil application. Available Online at:
372 <http://www.acadianseaplants.com/soilapplication.html>. Last Accessed: January 24, 2006.
- 373
- 374 Allen, V.G., K.R. Pond, K.E. Sakers, J.P. Fontenot, C.P. Bagley, R.L. Ivy, R.R. Evans, R.E. Schmidt, J.H. Fike,
375 X. Zhang, J.Y. Ayad, C.P. Brown, M.F. Millert, J.L. Montgomery, J. Mahan, D.B. Wester, C. Melton. 2001.
376 Tasco: Influence of a brown seaweed on antioxidants in forages and livestock – A review. *In J. Anim. Sci.*
377 79(E. Supple.): E21 – E31.
- 378
- 379 Baldwin, K. R. 2001. Soil fertility for organic farming. Available Online at:
380 http://www.ncsu.edu/organic_farming_systems/news/soil_fertility.PDF. Last Accessed: January 23,
381 2006.
- 382
- 383 Baker, B. 1996. Plant nutrition from the sea: Marine products can be used to supplement soil nutrients.
384 *Farmer to Farmer* 16: page numbers not listed. Available Online at:
385 <http://www.noamkelp.com/nutrition.html>. Last Accessed: January 23, 2006.
- 386
- 387 Canadian General Standards Board (CGSB). 2004. Organic Production Systems, Part 3 – Permitted
388 Substances lists -- DRAFT. Available Online at:
389 <http://www.cog.ca/documents/OrganicStandard2004.3.pdf>. Last Accessed: January 24, 2006.
- 390
- 391 Chalker-Scott, L. Date unknown. The myth of curative kelp: “Seaweed extracts reduce disease, improve
392 production, and increase stress resistance in landscape plants”. Available Online at:
393 [http://www.puyallup.wsu.edu/~Linda%20Chalker-](http://www.puyallup.wsu.edu/~Linda%20Chalker-Scott/Horticultural%20Myths_files/Myths/Seaweed%20extracts.pdf)
394 [Scott/Horticultural%20Myths_files/Myths/Seaweed%20extracts.pdf](http://www.puyallup.wsu.edu/~Linda%20Chalker-Scott/Horticultural%20Myths_files/Myths/Seaweed%20extracts.pdf). Last Accessed: January 31, 2006.
- 395
- 396 Conrad, Brad. July 16, 2002. Letter to Bob Pooler of the National Organics Standards Board from Arcadian
397 Sea Plants. .
- 398
- 399 Feenstra, G., C. Ingels, D. Campbell, D. Chaney, M.R. George, and E. Bradford. 1997. What is sustainable
400 agriculture? Available Online at: <http://www.sarep.ucdavis.edu/concept.htm>. Last Accessed: January 24,
401 2006.
- 402
- 403 Hall, B. 1997. Nonconventional Soil Amendments. *Appropriate Technology Transfer for Rural*
404 *Areas*. Fayetteville, AR (as cited in Baldwin, 2001).
- 405
- 406 Hall, B. and P. Sullivan. 2001. Alternative Soil Amendments. *Appropriate Technology Transfer for Rural*
407 *Areas*. Available Online at: <http://attra.ncat.org/attra-pub/PDF/altsoil.pdf>. Last Accessed: January 24,
408 2006.
- 409
- 410 Henry, Eric C. 2005. Report of Alkaline Extraction of Aquatic Plants. Available Online at:
411 http://www.omri.org/AdvisoryCouncil/Aquatic_plant_extract-2004-02-14.pdf. Last Accessed January
412 24, 2006.
- 413
- 414 International Food Information Council Foundation (IFIC). 1997. Everything you need to know about
415 glutamate and monosodium glutamate. Available Online at:
416 [http://ific.nisgroup.com/publications/brochures/upload/Everything-You-Need-To-Know-About-](http://ific.nisgroup.com/publications/brochures/upload/Everything-You-Need-To-Know-About-Glutamate-And-Monosodium-Glutamate.pdf)
417 [Glutamate-And-Monosodium-Glutamate.pdf](http://ific.nisgroup.com/publications/brochures/upload/Everything-You-Need-To-Know-About-Glutamate-And-Monosodium-Glutamate.pdf). Last Accessed: January 30, 2006.
- 418
- 419 Kuisma, P. 1989. The effect of foliar application of seaweed extract on potato. *Journal of Agricultural*
420 *Science in Finland* 61: 371-377 (as cited in Tourte 1997).

- 421
422 Laboratoires GOEMAR, S.A. undated. Technical document on Product GA 14. Provided as an attachment
423 to the Original TAP Database Form, 1995. (Fax dated from Agrimar Corp on January 12, 1995).
424
- 425 McGeary, D.J. and W.E. Birkenhead. 1984. Effect of Seaweed Extract on Growth and Yield of Onions. *The*
426 *Journal of the Australian Institute of Agricultural Science* 50: 49-50 (as cited in Tourte 1997).
427
- 428 North Central Regional's NCR-103 Committee on Nontraditional Soil Amendments and Growth
429 Stimulants (NCR). 2004. Non-conventional soil additives: products, companies, ingredients, and claims.
430 Available Online at: <http://www.soils.wisc.edu/extension/hottopics/nonconventional.pdf>. Last
431 Accessed: January 31, 2006.
432
- 433 Necessary Organics. 1991. Material Safety Data Sheet on SeaMix Liquid Fertilizer. Provided as an
434 attachment to the Original TAP Database Form, 1995.
435
- 436 Nelson, W.R. and J. Van Staden. 1984. The effect of seaweed concentrate on growth of nutrient-stressed
437 greenhouse cucumbers. *HortScience* 19(1): 81-82 (as cited in Tourte, 1997).
438
- 439 Nishimura, N., J. Zhang, M. Abo, A. Okubo, and S. Yamazaki. 2001. Simultaneous determination of betaines and
440 free amino acids in higher plants by capillary electrophoresis. *Analytical Sciences* 17(supplement): 929-932.
441
- 442 Organic Trade Association. 2002. Comparative Analysis of the United States National Organic Program (7
443 CFR 205) and the European Union Organic Legislation (EEC2092/91 & Amendments). Available Online at:
444 <http://www.ota.com/pics/documents/NOPEUunifiedreport.pdf>. Last Accessed: January 30, 2006.
445
- 446 OMRI 2005. OMRI Brand Name Product List. Organic Materials Review Institute. Eugene, Oregon.
447 January 2005.
448
- 449 Original TAP Review. 1999. Phosphoric Acid (Livestock). From NOSB Materials Database.
450
- 451 Original TAP Database Form. 1995. Aquatic Plant Extracts. From NOSB Materials Database.
452
- 453 Sideman, E. October 12, 2004. Letter to NOSB and NOP regarding comment on aquatic plant extracts.
454
- 455 Smittle, D. 1991. Seaweed comes ashore. *Fine Gardening Magazine*. Available Online at:
456 http://www.casoils.com/seaweed_comes_ashore.html. Last Accessed: January 23, 2006.
457
- 458 3R Lighting. 2005. Seaweed fertilizer. Available Online at:
459 <http://www.3rlighting.com/organic/fertilizer/seaweed.html>. Last Accessed: January 24, 2006.
460
- 461 Tooby, T. E. 2003. Registration opportunities for natural products versus synthetic plant stress signals (or
462 plant activators) for crop protection. Available Online at: http://www.ibma.ch/pdf/200311_paper.pdf.
463 Last Accessed: January 24, 2006.
464
- 465 Tourte, Laura. 1997. Kelp extract and fish powder sprays on organically grown processing tomatoes.
466 *Organic Farming Research Foundation's Information Bulletin* 4: 6-7 and 9.
467
- 468 U.S. Department of Agriculture's National Organic Program (NOP). 2005. National List: Regulatory Text.
469 Subpart G - Administrative. The National List of Allowed and Prohibited Substances. § 205.600 Evaluation
470 criteria for allowed and prohibited substances, methods, and ingredients. Available Online at:
471 <http://www.ams.usda.gov/nop/NOP/standards/ListReg.html>. Last Accessed: January 24, 2006.
472
- 473 U.S. Environmental Protection Agency (EPA). 1998. Certain plant regulators, cytokinins, auxins,
474 gibberellins, ethylene, and pelargonic acid; tolerance exemptions. *Federal Register* 63(205): 56882-56886.

475 Available Online at: <http://www.epa.gov/fedrgstr/EPA-PEST/1998/October/Day-23/p28360.htm>. Last
476 Accessed: January 31, 2006.
477
478 U.S. Geological Survey (USGS). 2001. Potash statistics and information. Available Online at:
479 <http://minerals.usgs.gov/minerals/pubs/commodity/potash/>. Last Accessed: January 30, 2006.
480
481 Verkleij, F.N. 1992. Seaweed extracts in agriculture and horticulture: a review. *Biological Agriculture and*
482 *Horticulture* 8: 309-324 (as cited in Tourte, 1997).
483
484